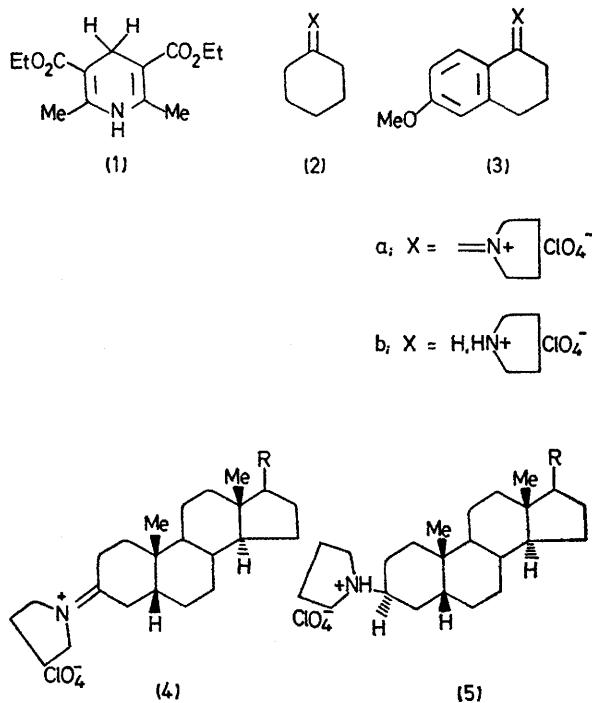


Reduction of the $>\overset{+}{\text{C}}=\overset{+}{\text{N}}<$ Bond by an NADH Model

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Summary Reduction of the iminium salts (**2a**), (**3a**), and (**4a—c**), with the Hantzsch ester (**1**), gives the corresponding ammonium salts, the reaction proceeding stereospecifically in the case of the steroidal systems (**4a—c**).

RECENTLY we communicated the stereospecific reduction of the carbon-carbon double bonds of $\alpha\beta$ -unsaturated iminium salts with 1,4-dihydropyridine derivatives.¹ We report



- a, R = C_8H_{17}
 b, R = OAc
 c, R = Ac

now the homogeneous reduction of the iminium chromophore, to the corresponding amine system, with the Hantzsch ester (1) (NADH model).^{2a,b} The iminium

† The salt (4c), m.p. 234–237 °C, was prepared analogously to (4a), m.p. 225 °C, and (4b), m.p. 245–247 °C. See ref. 1.

‡ Satisfactory spectral and microanalytical data have obtained for (5a–c); (5a), m.p. 281–283 °C; (5b), m.p. 270–271 °C; (5c) m.p. 266–267 °C.

§ ¹³C n.m.r. (CDCl_3), (5a): C-1 29.0, C-5 35.7; (5b): C-1 28.9, C-5 36.1; (5c): C-1 29.0, C-5 35.6 p.p.m. The chemical shifts for C-1 and C-5 are at a higher field (± 8.9 and ± 8.5 p.p.m., respectively) than in 5β -steroids without a C-3 substituent. C-3 α -substituents are known to cause only small diamagnetic shifts (1–2 p.p.m.) for C-1 and C-5. (D. Leibfritz and J. D. Roberts, *J. Amer. Chem. Soc.*, 1973, **95**, 4996). An example of a C-3 β -substituted 5β -steroid is reported to involve a high field shift of 7.9 p.p.m. at C-5. (G. Lukacs and C. R. Bennett in '¹³C N.m.r. Spectroscopy, Methods and Applications', Ed. E. Breitmaier and W. Voelter, Verlag Chemie, Weinheim 1974). A detailed discussion of the ¹³C n.m.r. spectra will be presented elsewhere.

¹ U. K. Pandit, F. R. Mas Cabré, R. A. Gase, and M. J. de Nie-Sarink, *J.C.S. Chem. Comm.*, 1974, 627.

² (a) A part of this work was communicated by us at the 9th Int. Symp. on the Chemistry of Natural Products, Ottawa, Canada, 1974. (b) For a related example see S. Shinkai and T. Kunitake, *Chem. Letters*, 1974, 1113.

³ U. K. Pandit, F. A. van der Vlugt, and A. C. van Dalen, *Tetrahedron Letters*, 1969, 3697.

perchlorates (2), m.p. 246–250 °C, and (3), m.p. 175 °C,³ were prepared from the corresponding enamines, while the salts (4a–c) were obtained by reduction of the conjugated iminium systems[†] which, in turn, were derived from the appropriate Δ^4 -3-keto-steroids. Reduction was carried out by refluxing the substrate iminium salt with compound (1) (1.2 equiv.) in MeCN until the starting iminium perchlorates could not be detected (i.r. absence of $\text{C}=\text{N}^+$ and/or t.l.c.). Evaporation and addition of ether gave the ammonium salts as crystalline products in excellent yields (> 90%). The structures of (2b) and (3b) were confirmed by liberation of the free amines (1M NaHCO_3) and their comparison with authentic samples prepared by NaBH_4 reduction of (2a), and (3a). While reduction of (4a–c) can, in principle, lead to isomeric steroidal amines, the product in each case was found to be a single compound. The gross structures of (5a–c) were deduced from their spectroanalytical data.‡ In view of the line-broadening of 3-H (and the pyrrolidine α -methylene hydrogens) due to quadrupole coupling, information on the stereochemistry of the pyrrolidine ring was derived from ¹³C n.m.r. spectra of the compounds. Particularly relevant to the stereochemical assignment of the C-3 substituent were the chemical shifts of C-1 and C-5, which, owing to the steric compression involving $\alpha\alpha'$ -pyrrolidinyll- and C-1 and C-5 (axial) protons, underwent an expected upfield shift.§ The stereochemical course of the reduction may be rationalized in terms of the sterically favoured α -approach of the reagent. However, this point requires further attention.

Pyridine nucleotide-mediated enzymatic reduction of the $>\text{C}=\text{N}$ - linkage is an important type of transformation in numerous biological processes. The facility with which iminium salts are reduced by NADH models strongly suggests that a protonated imine species may be involved in these reductions.

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